

Science Y11 Information Evening





Why do students study science?

THE SCIENCE WAY



We can make observations describe what we see We can explain everyday things in a scientific way We can team from successes 6 failures and adapt to do things better

We can explain everyday things in a scientific way with people with people with different skills & knowledge

WE EVALUATE EXPERIMENTAL RESULTS IN LIGHT OF THE ORIGINAL PROBLE

We use scientific vocabulary accurately & talk like a

We can use numbers and data to support our work and obtain meaningful information

We can identify key issues in a problem and use our scientific knowledge to tackle them

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Studying GCSE Science is essential for:

- Career opportunities
- Higher education
- Critical thinking, understanding the world
- Future-proofing your education.



SUBJECT WAYS



Exam Dates

6 exam papers altogether: 2x biology, 2x chemistry, 2 x physics. Each paper is 1h15mins (1hr45mins for separate).

	Biology	Chemistry	Physics
Paper 1	12 th May	20 th May	2nd June
Paper 2	8 th June	12 th June	15 th June



How is science assessed?

Combined Foundation

Combined Higher

Separate Higher

Combined grades,
 11-55 (set 4-7)

 Combined grades, 44-99 (set 2 and 3)

 Separate grades for each science, 4-9 (set 1s)

Grades are worked out on an average of all their science papers

6 exam papers altogether: 2x biology, 2x chemistry, 2 x physics. Each paper is 1h15mins (1hr45mins for separate).



What's on each exam paper?

Biology

- <u>Paper 1:</u>
- B1 Cell biology
- B2 Organisation
- B3 Infection & response
- B4 Bioenergetics
- Paper 2:
- B5 Homeostasis & response
- B6 Inheritance, variation & evolution
- B7 Ecology

Chemistry

- <u>Paper 1:</u>
- C1 Atomic structure & periodic table
- C2 Bonding & structure
- C3 Quantitative chemistry
- C4 Chemical changes
- C5 Energy of changes
- **Paper 2**:
- C6 Rate and extent of reactions
- C7 Organic chemistry
- C8 Chemical analysis
- C9 Earth and atmosphere
- C10 Using resources

Physics

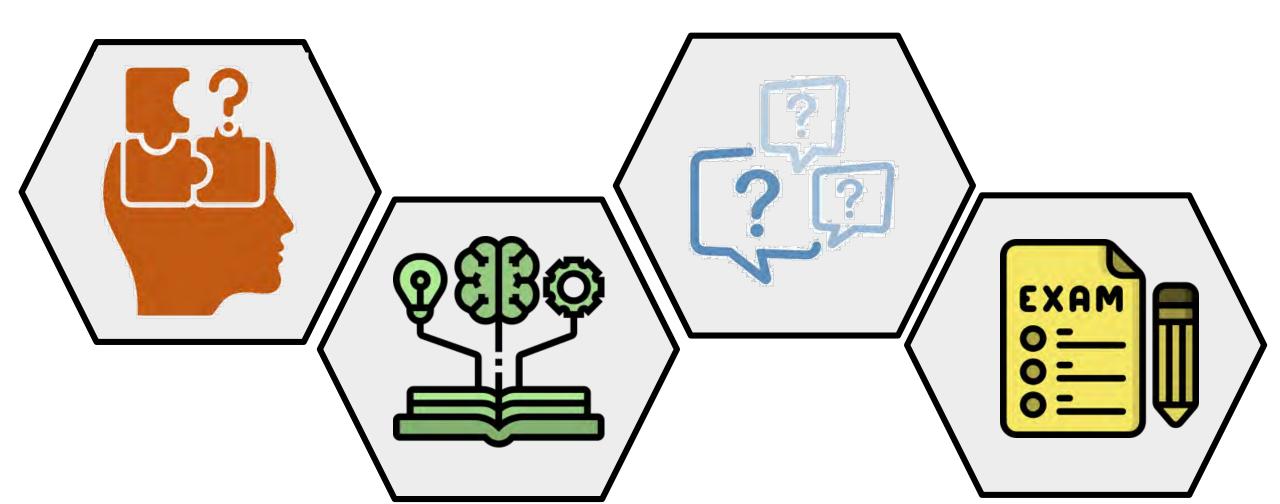
- <u>Paper 1:</u>
- P1 Energy
- P2 Electricity
- P3 Particle model of matter
- P4 Atomic structure
- Paper 2:
- P5 Forces
- P6 Waves
- P7 Magnetism
- P8 Space* (separate only)



Independent Study Program

Step 1
Identify Gaps

Step 2 Relearn Step 3 Recall Step 4
Practice





Step 1 Identifying Gaps



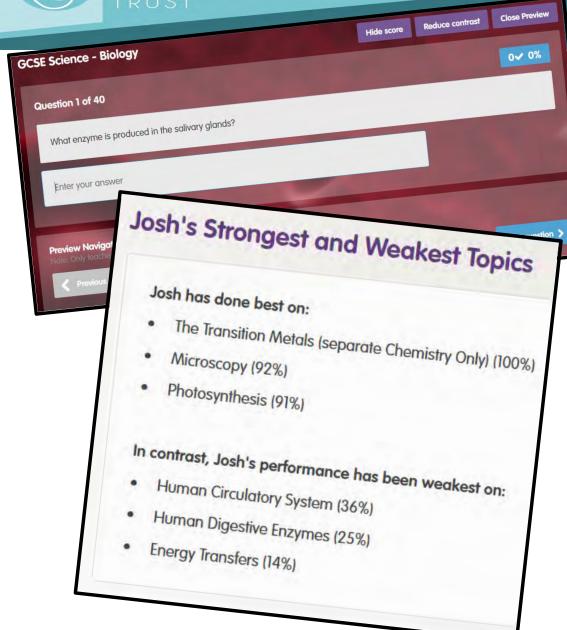
This step is crucial because it pinpoints areas where understanding is weak, allowing students to focus revision efforts effectively.



KNOW	more REMEMBER more	
GCSE Physics: Foundation/Higher/S Review the checkpoints below. Y practice them until you do.	P2 - Electricity ou should make flashcards on any point that you don't feel confident on and	
□ The different circuit symbols: □ ○ witch (open) □ ○ switch (closed) □ □ cell □ □ battery	lamp diode fuse resistor V voltmeter variable resistor LDR	
For current to flow, there must be a closed circuit and a source of potential difference (battery/cell). Electric current is the rate of flow of charge. Current is measured using an ammeter connected in series. Potential difference is measured using a voltmeter connected in parallel. Charge is measured in coulombs. Current is measured in amps. Potential difference is measured in volts. Resistance is measured in ohms.		

r paper to calculate the marks awarded for the number of marks you achieved for ead column tells you the total number of marks.	Question	Marks Available	abou	teel at Your gress in Topic?	
mic Structure and the Periodic Table	02.1 02.2 02.3	02.4 11	II	F	
mic Structure and the	01.2 01.3 01.4	01.5 9			
-:- Christure	01.2 01.3			TT	
e Periodic Table onding, Structure and the Properties of	04.1 05.1	2	++	++	7
hemical Bonds Now Bonding and Structure are Related to he Properties of Substances Structure and Bonding of Carbon	01.1 04.3		4		3
Quantitative Chemical Substances Chemical Measurements, Conservation of Mass and Chemical Equations Use of Amount of Substance in Relation Masses of Pure Substances	10	05.2	8		
chemical Changes	03.1 05.3	07.7	6	++	+
Peactivity of Metals	07.4 07.	5 03.6	6		
Reactions of Acids	03.4 03				T
Electrolysis	1-71 0	72 07.3 0	7.5 8		_
Eyothermic and Endourier	0(15				
Mass and Chemical Equation Use of Amount of Substance in Relation: Masses of Pure Substances Chemical Changes Reactivity of Metals Reactions of Acids Electrolysis Energy Changes Exothermic and Endothermic Reaction Which topics did you do really well in	03.1 05.3 07.4 07. 03.4 03	6 07.7	6		H







Teachers can set quizzes, but students can set their own quizzes themselves.

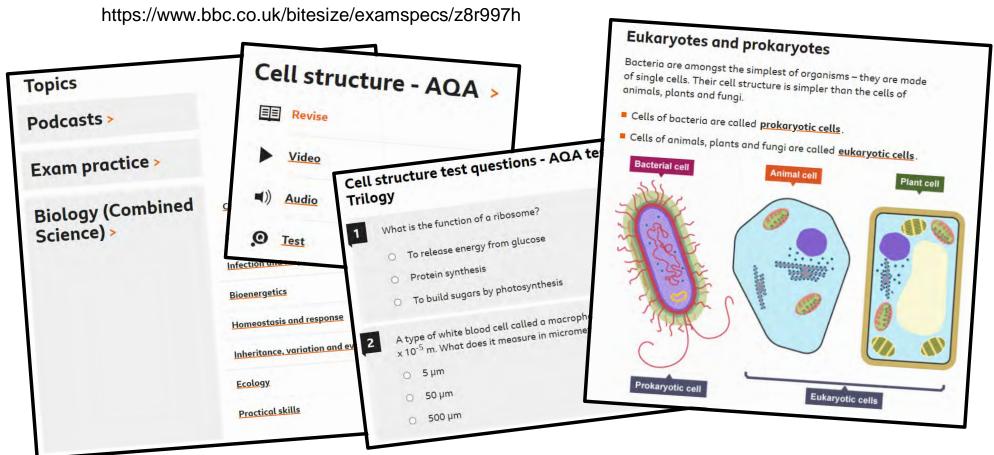
Tests mark themselves so they get immediate feedback and strengths & targets.

	Percentage Correct	Questions Answered
Biology	69%	633
Chemistry	65%	504
Physics	62%	365
Maths for Science	0%	0
Norking Scientifically	0%	0





BBC bitesize



Similar to an online revision guide, organised in topic order with clear information and videos to help understand a topic.

Ensure your child is on the correct revision page, i.e. AQA combined science.



Revision guides

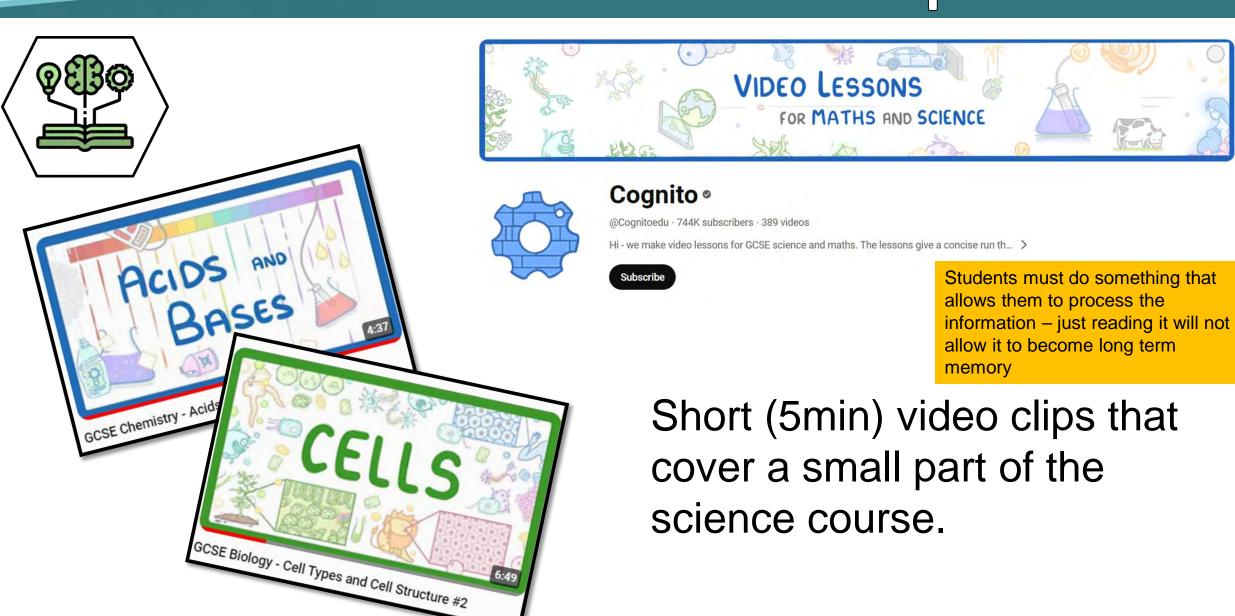


Revision Biology (Separate) - BAR46	3.30
Revision Chemistry (Separate) - CAR46	3.30
Revision Physics (Separate) - PAR48	3.30
Workbook Biology (Separate) - BAQ42	3.30
Workbook Chemistry (Separate) - CAQ42	3.30
Workbook Physics (Separate) - PAQ42	3.30
Biology Revision Cards (Separate) - BAF41	5.80
Chemistry Revision Cards (separate) - CAF41	5.80
Physics Revision Cards (Separate)- PAF41	5.80



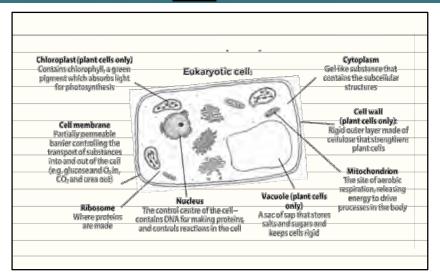
Revision Guide Combined Higher - SAHR46	6.35
Workbook Combined Higher - SAHQ42	6.35
Revision Guide Combined Foundation - SAFR46	6.35
Workbook Combined Foundation - SAFQ42	6.35
Question Cards - All-in-one(for Combined Higher and Foundation) - SCAF41	10.35







Front



Back

- List three structures found only in plant cells, but not animal cells
- Explain the difference between passive transport and active transport across the cell membrane.
- 3. Where in a eukaryotic cell does cellular respiration occur, and what are the key products of this process

If you use this technique, it is important to remember that the creation of the flashcards does not represent the end point of revision for this topic.

The next crucial step is to test how well you can recall the information on the flashcard (Retrieval).

Top Tips

Pick one concept per card:

Students must do something that allows them to process the information – just reading it will not allow it to become long term memory

- Questions on one side, answers/information on the other
- Keep it concise: Focus on key terms and important details.
- Draw diagrams and use colour. (dual coding)

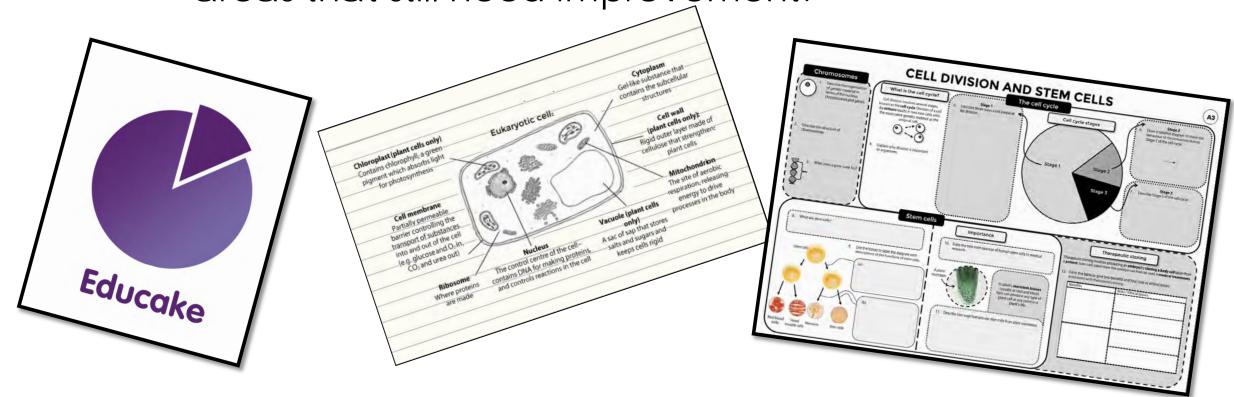


Step 3 Recall



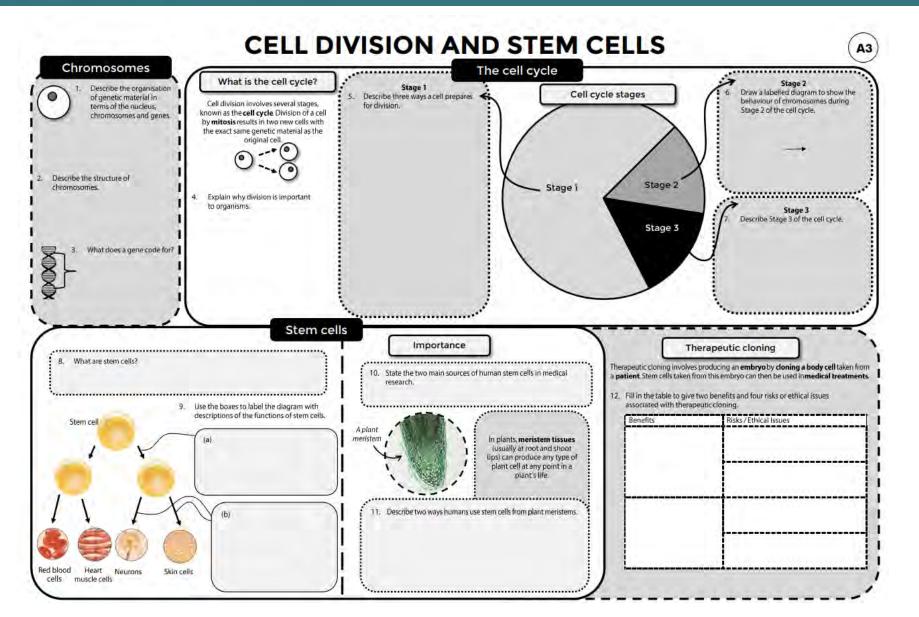
This stage should be testing that you have learnt the new knowledge.

This step helps consolidate understanding and identify areas that still need improvement.





Step 3 Recall



Try and complete retrieval grids from *memory*.

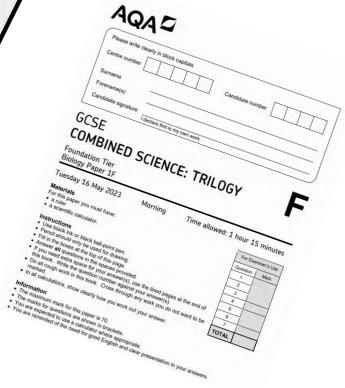
Revisit any gaps in knowledge before moving on to exam style questions

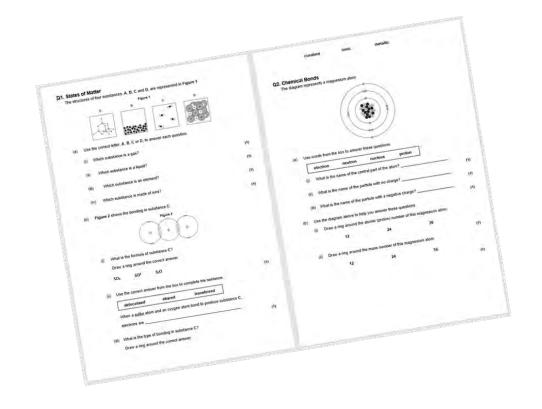


Step 4 Practice



Once secure in the knowledge and understanding it is time to start applying this to exam questions







How can you help as a parent?

- No distractions (no phones, no TV, no headphones)
- Help them plan when/how to revise little and often
- Quiz them using their Q&A flashcards
- If they are struggling encourage them to keep going and if they are really stuck ask the teacher
- Encourage them to use exam questions and mark schemes
 Challenge yourself if its too easy its
 probably not effective!